

Number	Hits	Search Text	DB	Time stamp
	0	("wc-9944972-\$ did").PN.	EPO; JPO; DERWENT	2002/06/23 14:13
	1	wo-9944972-\$ did.	EPO; JPO; DERWENT	2002/06/23 14:46
2	235	252/183.11-183.12.ccls.	USPAT; US-PPGPUB	2002/06/23 14:47
3	1233	n,n-diethylhydroxylamine	USPAT; US-PPGPUB	2002/06/23 14:47
4	0	252/183.11-183.12.ccls. and n,n-diethylhydroxylamine	USPAT; US-PPGPUB	2002/06/23 14:47
5	2208	diethylhydroxylamine	USPAT; US-PPGPUB	2002/06/23 14:47
6	341	phenylhydroxylamine	USPAT; US-PPGPUB	2002/06/23 14:47
7	0	252/183.11-183.12.ccls. and diethylhydroxylamine and phenylhydroxylamine	USPAT; US-PPGPUB	2002/06/23 14:48
8	0	252/183.11-183.12.ccls. and phenylhydroxylamine	USPAT; US-PPGPUB	2002/06/23 14:55
9	2	252/183.11-183.12.ccls. and diethylhydroxylamine	USPAT; US-PPGPUB	2002/06/23 14:55
10	0	252/182.29,403,404.ccls.	USPAT; US-PPGPUB	2002/06/23 14:55
11	2346	252/182.29,403,404.ccls.	USPAT; US-PPGPUB	2002/06/23 14:55
12	4	phenylhydroxylamine and 252/182.29,403,404.ccls.	USPAT; US-PPGPUB	2002/06/23 15:00
13	30	"n-nitroso-n-phenylhydroxylamine"	USPAT; US-PPGPUB	2002/06/23 15:04
14	15368	polymerization same inhibit\$4	USPAT; US-PPGPUB	2002/06/23 15:05
15	245	diethylhydroxylamine and (polymerization same inhibit\$4)	USPAT; US-PPGPUB	2002/06/23 15:06
16	5	(diethylhydroxylamine and (polymerization same inhibit\$4)) and phenylhydroxylamine	USPAT; US-PPGPUB	2002/06/23 15:06
17	2821	polyester and bromine and antimony	USPAT; US-PPGPUB	2002/06/23 17:15
18	23460	flame ADJ10 retard\$4	USPAT; US-PPGPUB	2002/06/23 17:16
19	1498	((polyester and bromine and antimony) and (flame ADJ10 retard\$4))	USPAT; US-PPGPUB	2002/06/23 17:16
20	19717	ptfe or polytetrafluoroethylnene	USPAT; US-PPGPUB	2002/06/23 17:16
21	90	((polyester and bromine and antimony) and (flame ADJ10 retard\$4)) and (ptfe or polytetrafluoroethylnene)	USPAT; US-PPGPUB	2002/06/23 08:19
22	801	219/121.69.ccls.	USPAT; US-PPGPUB	2002/06/28 17:50
23	0	((polyester and bromine and antimony) and (flame ADJ10 retard\$4)) and (ptfe or polytetrafluoroethylnene)) and 219/121.69.ccls.	USPAT; US-PPGPUB	2002/06/28 17:46
24	68326	219/\$.ccls.	USPAT; US-PPGPUB	2002/06/28 17:46
25	0	((polyester and bromine and antimony) and (flame ADJ10 retard\$4)) and (ptfe or polytetrafluoroethylnene)) and 219/\$.ccls.	USPAT; US-PPGPUB	2002/06/28 17:46
26	2	((polyester and bromine and antimony) and (flame ADJ10 retard\$4)) and 219/\$.ccls.	USPAT; US-PPGPUB	2002/06/28 17:50
27	3524	219/121.6-121.69.ccls.	USPAT; US-PPGPUB	2002/06/28 17:50
28	210130	laser	USPAT; US-PPGPUB	2002/06/28 17:51
29	3269	219/121.6-121.69.ccls. and laser	USPAT; US-PPGPUB	2002/06/28 17:51
30	655739	mark\$4	USPAT; US-PPGPUB	2002/06/28 17:51

914	((219/121.6-121.69.cccls. and laser) and mark\$4	USPAT; US-PGPUB	2002/06/28 17:52
0	((polyester and bromine and antimony) and (flame ADJ10 retard\$4)) and ((219/121.6-121.69.cccls. and laser) and mark\$4)	USPAT; US-PGPUB	2002/06/28 17:53
0	((polyester and bromine and antimony) and (flame ADJ10 retard\$4)) and ((219/121.6-121.69.cccls. and laser))	USPAT; US-PGPUB	2002/06/28 17:53
52	((219/121.6-121.69.cccls. and laser) and mark\$4) and polyester	USPAT; US-PGPUB	2002/06/28 18:02
64918	terephthalate	USPAT; US-PGPUB	2002/06/28 18:02
26	((219/121.6-121.69.cccls. and laser) and mark\$4) and terephthalate	USPAT; US-PGPUB	2002/06/28 18:02
1	5783105.pn.	USPAT; US-PGPUB	2002/06/29 08:47
50	(44/597).CCLS.	USPAT; US-PGPUB	2002/06/29 08:54
49	(44/553).CCLS.	USPAT; US-PGPUB	2002/06/29 14:11

DN 113:232.147
 TI Epoxidation of cyclohexenylmethyl (meth)acrylate in presence of
 polymerization inhibitors
 IN Fukuya, Kazuaki; Kuwana, Akihiro
 PA Daicel Chemical Industries, Ltd., Japan
 SO Jpn. Kokai Tokkyo Koho, 9 pp.
 CDDRN: JKXXAF
 DT Patent
 LA Japanese
 IC ICM C07B301-35
 ICS C07B303-16; C08F010-32; C08G059-20; C09D004-02; C09D133-06;
 C09D163-00
 CC 35-2 (Chemistry of Synthetic High Polymers)
 Section cross-reference(s): 21
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	-----	-----	-----	-----
JP 00188576	A2	19900724	JP 1989-5816	19890112
JP 27041984	B2	19980126		

OS MARPAT 113:232247
 AB 3,4-Epoxycyclohexylmethyl acrylate and methacrylate (I) are prepd. by
 epoxidn. of 3-cyclohexen-1-ylmethyl acrylate and methacrylate (II) with
 an
 oxidizing agent in the presence of polymn. inhibitors comprising
 .gtoreq.1
 compd. selected from hydroquinone, hydroquinone mono-Me ether (III),
 p-benzoquinone, cresol, tert-butylcat•chol, phenols substituted by
 tert-Bu
 and other groups, 2,5-dihydroxy-p-quinone, piperidine, ethanolamine,
 .alpha.-nitroso-.beta.-naphthol, HNPh₂, phenothiazine,
 N-nitrosophenylhydroxylamine, and Et₂NOH and .gtoreq.1 compd. selected
 from H₃PO₄, K₃PO₄, Na₃PO₄, Na(NH₄)HPO₄, H₄P₂O₇, K₄P₂O₇, Na₄P₂O₇,
 2-ethylhexyl pyrophosphate, K or Na 2-ethylhexyl pyrophosphate,
 tripolyphosphoric acid, K or Na 2-ethylhexyl tripolyphosphate, and Na or
 K
 2-ethylhexyl tetrapolyphosphate. Thus, a mixt. of 14.4 kg II, 52.8 kg
 AcOEt, 12 g III, and 12 g H₄P₂O₇ was treated with 24.8 kg 30% AcOOH
 during
 4 h at 50°. and aged 4 h to give 14.2 kg product contg. 94.7% I, 1
 g
 of which dissolved completely in 10 g heptane.
 ST epoxycyclohexylmethyl acrylate prepn polymn inhibitor; methacrylate
 epoxycyclohexylmethyl prepn polymn inhibitor; epoxidn cyclohexenylmethyl
 acrylate polymn inhibitor; hydroquinone polymn inhibitor methacrylate;
 pyrophosphoric polymn inhibitor acrylate; phenol polymn inhibitor
 acrylate; amine polymn inhibitor acrylate; phosphate polymn inhibitor
 acrylate
 IT Polymerisation inhibitors
 (in epoxidn. of cyclohexenylmethyl (meth)acrylate)
 IT Epoxidation
 (of cyclohexenylmethyl (meth)acrylate, polymn. inhibitors in)
 IT Phenols, uses and miscellaneous
 RL: MFRS (Uses)
 (polymn. inhibitors, in epoxidn. of cyclohexenylmethyl (meth)acrylate)
 IT 21367-03-3, 3-Cyclohexen-1-ylmethyl acrylate 21367-03-3,
 3-Cyclohexen-1-ylmethyl methacrylate
 RL: ECT (Reactant)
 (epoxidn. of, polymn. inhibitors in)

IT 38-32-4, 3-tert-Butyl-4-methoxyphenol 92-84-2, Phenothiazine
106-51-4,

p-Benzoquinone, uses and miscellaneous 110-89-4, Piperidine, uses and
miscellaneous 121-00-6, 2-tert-Butyl-4-methoxyphenol 122-39-4,
Diphenylamine, uses and miscellaneous 123-31-9, Hydroquinone, uses and
miscellaneous 128-37-0, 2,6-Di-tert-butyl-p-cresol, uses and
miscellaneous 131-91-9, .alpha.-Nitroso-.beta.-naphthol 141-43-5,
Ethanolamine, uses and miscellaneous **148-97-0** 150-76-5,
Hydroquinone monomethyl ether 615-94-1 1319-77-3, Cresol 1693-78-3,
2-Ethylhexyl pyrophosphate 1879-09-0, 2,4-Dimethyl-6-tert-butylphenol
1466-09-3, Fyrcphosphoric acid **3710-84-7**, N,N-
Diethylhydroxylamine 7320-34-5, Potassium pyrophosphate 7632-05-5,
Sodium phosphate 7664-38-2, Phosphoric acid, uses and miscellaneous
7722-88-5 10380-03-2, Tripolyphosphoric acid 12767-83-8, Sodium
2-ethylhexyl tripolyphosphate 13011-54-6, Ammonium sodium hydrogen
phosphate 16068-46-5, Potassium phosphate 27213-78-1,
tert-Butylcatechol 130455-01-5 130455-02-6 130455-03-7

130455-65-1

130455-66-2

RL: USES (Uses)

(polymn. inhibitors, in epoxidn. of cyclohexenylmethyl (meth)acrylate)

IT 64630-63-3P, 3,4-Epoxycyclohexylmethyl acrylate 82428-30-6P,
3,4-Epoxycyclohexylmethyl methacrylate

RL: PREP (Preparation)

(prepn. of, by epoxidn., polymn. inhibitors in)

L2 ANSWER 33 OF 44 CAPLUS COPYRIGHT 2002 ACS

AN 1966:457147 CAPLUS

DN 65:57147

OREF 65:10697f,10698a

TI Polymerization inhibitors

EA Copolymer Rubber and Chemical Corp.

SD 12 pp.

DT Patent

LA Unavailable

IC 008F

CC 45 (Synthetic High Polymers)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	NL 65011747		19660311	NL	
PRAI	US		19640910		
AB	Org. N-nitrosohydroxylamines or salts thereof are used in vinylpyridine, unsatd. hydrocarbons, and unsatd. esters to inhibit thermal polymerization and (or) the growth of popcorn polymer therein, e.g. during storage.				

They

need not be removed prior to use of the monomers in catalytic polymerizations. NH₄ salts of **N-nitroso-N-phenylhydroxylamine** (I) or of N-nitroso-N-(1-naphthyl)hydroxylamine are very useful. I is twice as effective as a thermal polymerization inhibitor than the conventional *tert*-butylpyrocatechol (II). Copolymerization of butadiene with styrene by a cold rubber polymerization process is not retarded by I while II prevents any reaction.

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FILE COVERS 1907 - 29 Jun 2002 VOL 137 ISS 1
FILE LAST UPDATED: 27 Jun 2002 (20020627/ED)

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```
= s n-nitros-n-phenylhydroxylamine
2436803 N
    57 NITROS
    34 NITROSES
    91 NITROS
        (NITROS OR NITROSES)

2436803 N
    1596 PHENYLHYDROXYLAMINE
    198 PHENYLHYDROXYLAMINES
    1685 PHENYLHYDROXYLAMINE
        (PHENYLHYDROXYLAMINE OR PHENYLHYDROXYLAMINES)
LI      0 N-NITROS-N-PHENYLHYDROXYLAMINE
        (N(W)NITROS(W)N(W)PHENYLHYDROXYLAMINE)

= s n-nitroso-n-phenylhydroxylamine
2436803 N
    22271 NITROSO
    15 NITROSOS
    22276 NITROSO
        (NITROSO OR NITROSOS)

2436803 N
    1596 PHENYLHYDROXYLAMINE
    198 PHENYLHYDROXYLAMINES
    1685 PHENYLHYDROXYLAMINE
        (PHENYLHYDROXYLAMINE OR PHENYLHYDROXYLAMINES)
LC      44 N-NITROSO-N-PHENYLHYDROXYLAMINE
        (N(W)NITROSO(W)N(W)PHENYLHYDROXYLAMINE)

= s 100-65-2/rn
    830 100-65-2
```

L3 30 100-65-2D
L3 803 100-65-2//RN
(100-65-2 (NOTL) 100-65-2D)

= s 12 and 13
L4 0 L2 AND L3

= d 12 1-44 ti

L2 ANSWER 1 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Determination of bismuth, selenium and tellurium in nickel-based alloys and pure copper by flow-injection hydride generation atomic absorption spectrometry with ascorbic acid pre-reduction and cupferron chelation-extraction

L2 ANSWER 2 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Stabilized monomer composition

L2 ANSWER 3 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Separation and direct UV detection of lanthanides complexed with cupferron by capillary electrophoresis

L2 ANSWER 4 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI O-Alkylation of Cupferron: Aiming at the Design and Synthesis of Controlled Nitric Oxide Releasing Agents

L2 ANSWER 5 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Polymerization inhibiting compositions, noncorrosive inhibitors and method for using

L2 ANSWER 6 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Unsaturated polyester-based coating compositions containing N-substituted N-nitrosohydroxylamine salts

L2 ANSWER 7 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Synthesis and spectral data of some new **N-nitroso-N-phenylhydroxylamine** (cupferron) derivatives

L2 ANSWER 8 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Determination of trace europium by adsorptive cathodic stripping voltammetry after complexation with cupferron

L2 ANSWER 9 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Nitroso compounds for use as antioxidants for preparation of (meth)acrylate esters bearing alicyclic epoxy groups

L2 ANSWER 10 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Preparation of N,N-dimethylacrylamide or N,N-dimethylmethacrylamide

L2 ANSWER 11 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Purifying aqueous solutions of indium salts

L2 ANSWER 12 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Purification of unsaturated carboxylic acid isocyanatoalkyl esters by distillation

L2 ANSWER 13 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Discharge characteristics of metal complexes of N-nitroso-N-phenylhydroxylamine as cathode materials for lithium primary cells

L2 ANSWER 14 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Extraction-polarographic determination of trace metals in rubbers

L2 ANSWER 15 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Polymerization inhibitors for acrylic monomers

L2 ANSWER 16 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Selective removal of trace copper ion in nickel electroplating bath with chelating reagents

L2 ANSWER 17 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI In vivo and in vitro inhibition of mung bean superoxide dismutase by cupferron

L2 ANSWER 18 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Azo- and azoxy compounds. IV. Alkylation of **N-nitroso-N-phenylhydroxylamine**. Synthesis and mass spectra of 1-alkoxydiazene 2-oxides

L2 ANSWER 19 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Photocuring compositions

L2 ANSWER 20 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Reaction of **N-nitroso-N-phenylhydroxylamine** with epoxides

L2 ANSWER 21 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Photocurable coating materials

L2 ANSWER 22 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Photocurable coating materials

L2 ANSWER 23 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Planar dicarbonylrhodium(I) and -iridium(I) complexes with polarizable aromatic ligands

L2 ANSWER 24 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Ultraviolet and infrared spectra of tetrakis(cupferronato) and (neocupferronato)uranium(IV)

L2 ANSWER 25 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Light-sensitive photographic material

L2 ANSWER 26 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Polymerization inhibitors for the catalytic hydration of acrylonitrile

L2 ANSWER 27 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Ultraviolet and infrared spectra of cupferron and neocupferron

L2 ANSWER 28 OF 44 CAPLUS COPYRIGHT 2002 ACS
TI Thermodynamics of metal-ligand bond formation. III. Adducts of heterocyclic bases with bis(N-nitroso-N-phenylhydroxylaminato)copper(II)

L2 ANSWER 29 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Preconcentration of trace amounts of molybdenum in soil extract by coprecipitation with cupferron in the presence of iron

L2 ANSWER 30 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Adsorption of iodide by soils

L2 ANSWER 31 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Some complexes of americium and curium with oxine, cupferron, and N-benzoylphenylhydroxylamine

L2 ANSWER 32 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Determination of vanadium by atomic absorption spectrophotometry

L2 ANSWER 33 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Polymerization inhibitors

L2 ANSWER 34 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI New heterocycles containing boron and nitrogen

L2 ANSWER 35 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Coordination compounds of pentavalent vanadium

L2 ANSWER 36 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI New synthesis of N-nitroso-N-arylhydroxylamines

L2 ANSWER 37 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI The distribution coefficient of cupferron

L2 ANSWER 38 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Fungicidal solubilized metal salts of N-nitroso-N-arylhydroxylamines

L2 ANSWER 39 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Extraction and flame-spectrophotometric determination of vanadium

L2 ANSWER 40 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Effect of chelating agents on the survival of irradiated mice

L2 ANSWER 41 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Stabilization of polymerizable heterocyclic nitrogen compounds

L2 ANSWER 42 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Choline dehydrogenase of the liver

L2 ANSWER 43 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI Salts of **N-nitroso-N-phenylhydroxylamines** as fungicides and bactericides

L2 ANSWER 44 OF 44 CAPLUS COPYRIGHT 2002 ACS

TI The extraction of metal complexes. IV. The dissociation constants and partition coefficients of 8-quinolinol (oxine) and N-nitro-N-phenylhydroxylamine (cupferron)

=> d 12 2 5 6 10 15 26 33 all

L2 ANSWER 1 OF 44 CAPLUS COPYRIGHT 2002 ACS
AN 2001:152347 CAPLUS
DN 134:193861

TI Stabilized monomer composition
 IN Schärf, Jakob; Rau, Hartmut; Gotzen, Friedrich
 PA Fohm GmbH, Germany
 SO Eur. Pat. Appl., 8 pp.
 CODEN: EPXXIW
 DT Patent
 LA German
 IC ICM C09K019-20
 ICS C07C007-20; C07B063-04
 CC 36-2 (Chemistry of Synthetic High Polymers)
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP 1078973	A2	20010228	EP 2000-117696	20000817
	EP 1078973	A3	20010307		
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO				
	DE 19940623	A1	20010301	DE 1999-19940623	19990827
	JP 2001089417	A2	20010403	JP 2000-254164	20000824
PRAI	DE 1999-19940623	A	19990827		
AB	A storage-stable synergistic inhibitor compn., useful esp. in the manuf. of hydroxalkyl (meth)acrylate monomers, comprises (1-10):1 (wt. parts) mixt. of Et ₂ NOH and PhN(NO)OH, resp., as synergistic inhibitor combination.				
ST	polymn inhibitor diethylhydroxylamine phenylnitrosohydroxylamine; nitrosophenylhydroxylamine diethylhydroxylamine polymn inhibitor hydroxyethyl acrylate; hydroxyethyl acrylate manuf stabilization diethylhydroxylamine phenylnitrosohydroxylamine synergistic inhibitor				
IT	Monomers				
	FL: MSC (Miscellaneous) (ethylenically unsatd.; stabilized monomer compn. contg. synergistic combination of N,N-diethylhydroxylamine and N-nitroso-N-phenylhydroxylamine)				
IT	Polymerization inhibitors (stabilized monomer compn. contg. synergistic combination of N,N-diethylhydroxylamine and N-nitroso-N-phenylhydroxylamine)				
IT	148-97-0, N-Nitroso-N-phenylhydroxylamine FL: NUU (Other use, unclassified); USES (Uses) (stabilized monomer compn. contg. synergistic combination of N,N-diethylhydroxylamine and)				
IT	818-61-1, 2-Hydroxyethyl acrylate 5205-93-6 13081-44-2, N,N-Dimethylaminoethyl methacrylamide FL: MSC (Miscellaneous) (stabilized monomer compn. contg. synergistic combination of N,N-diethylhydroxylamine and N-nitroso-N-phenylhydroxylamine)				
IT	135-20-6, Cupferron FL: NUU (Other use, unclassified); USES (Uses) (stabilized monomer compn. contg. synergistic combination of N,N-diethylhydroxylamine and N-nitroso-N-phenylhydroxylamine)				
IT	3710-84-7, N,N-Diethylhydroxylamine FL: NUU (Other use, unclassified); USES (Uses) (stabilized monomer compn. contg. synergistic combination of N-nitroso-N-phenylhydroxylamine and)				

L2 ANSWER 5 CF 44 CAPLUS COPYRIGHT 2002 ACS
AN 1999-176891 CAPLUS
DN 131:20.0775
TI Polymerization inhibiting compositions, noncorrosive inhibitors and
method

for using
IN UKita, Keizo; Onodera, Yuko
PA Nippon Zeon Co., Ltd., Japan
SO PCT Int. Appl., 64 pp.
CODEN: PIIXDE
DT Patent
LA Japanese
IC ICM C07C011-18
ICS C07C011-167; C07C239-08; C07C007-20; C07F009-09; C07F009-50;
C08F002-40
CC 37-2 (Plastics Manufacture and Processing)
FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 9944972	A1	19990910	WO 1999-JP1017	19990303
	W: CN, ID, JP, KR, US PW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE				
	EP 1061059	A1	20001220	EP 1999-907850	19990303
	R: DE, ES, FR, GB, IT, NL, PT				

PRAI JP 1998-67872 A 19980303
JP 1998-292987 A 19980930
WO 1999-JP1017 W 19990303

OS MARPAT 131:200775
AB The inhibitor compns. comprise (A) .gtoreq.1 compd. having a NO radical
group or its precursor and (B) a P-contg. compd. as corrosion inhibitor
in an A/B wt. ratio of 1:10 to 100:1. Monomer compns. contg. conjugated
dienes, arom. vinyl compds., ethylenically unsatd. nitrile compds. or/and
.alpha.-olefins are effectively inhibited from polymn. by including the
inhibitors during handling and storage. Thus, heating a compn. of 20 g
conjugated diene-contg. hydrocarbon mixt. in the presence of 180 ppm
(added at 60 ppm over 8 h for 3 times), Fe flakes,

N,N-diethylhydroxylamine
(480 ppm over 8 h for 3 times) and Latemul P 909 (phosphate based
surfactant; 480 ppm over 8 h for 3 times) for 24 h at 125.degree. showed
polymer formation 0.07%, high boiling fraction 0.25% and no corrosion of
Fe flakes.

ST anticorrosive phosphate surfactant polymn inhibitor nitroxide; diene
monomer polymn inhibitor nitroxide radical

IT Alkadienes
PL: RCT (Reactant); RACT (Reactant or reagent)
(conjugated, monomers; polymn. inhibiting compns., noncorrosive
inhibitors and method for using)

IT Polymerization inhibitors
(hindered nitroxides; polymn. inhibiting compns., noncorrosive
inhibitors and method for using)

IT Alkadienes
PL: MSC (Miscellaneous)
(monomers; polymn. inhibiting compns., noncorrosive inhibitors and
method for using)

IT Vinyl compounds, reactions
PL: PCT (Reactant); RACT (Reactant or reagent)

(monomers; polymn. inhibiting compns., noncorrosive inhibitors and method for using)

IT Corrosion inhibitors (polymn. inhibiting compns., noncorrosive inhibitors and method for using)

IT Nitroxides

FL: MOA (Modifier or additive use); USES (Uses) (polymn. inhibitors/precursors; polymn. inhibiting compns., noncorrosive inhibitors and method for using)

IT 2516-92-9, Bis(1-oxyl-2,2,6,6-tetramethylpiperidine-4-yl) sebacate

RL: MOA (Modifier or additive use); USES (Uses) (Ciba 5415, polymn. inhibitor precursors; polymn. inhibiting compns., noncorrosive inhibitors and method for using)

IT 121-45-9 554-70-1, Triethylphosphine 603-35-0, Triphenylphosphine, uses 1608-26-0 3049-24-9, Triphenyl phosphonate 7558-79-4, Disodium phosphate 7558-80-7, Monosodium phosphate 7664-38-2, Phosphoric acid, uses 7664-38-3D, Phosphoric acid, esters or salts, uses 9021-89-0 9071-85-6 26523-78-4, Tris(nonylphenyl) phosphite 51811-79-1, Latemul P 909 82905-49-5, Pelex RP

FL: MOA (Modifier or additive use); USES (Uses) (corrosion inhibitors; polymn. inhibiting compns., noncorrosive inhibitors and method for using)

IT 78-79-5, reactions 106-99-0, 1,3-Butadiene, reactions

RL: RCT (Reactant); RACT (Reactant or reagent) (monomers; polymn. inhibiting compns., noncorrosive inhibitors and method for using)

IT 100-42-5, reactions

RL: PCT (Reactant); RACT (Reactant or reagent) (polymn. inhibiting compns., noncorrosive inhibitors and method for using)

IT 135-20-6, **N-Nitroso-N-phenylhydroxylamine** ammonium salt 3710-84-7 7632-00-0, Sodium nitrite

RL: MOA (Modifier or additive use); USES (Uses) (polymn. inhibitor precursors; polymn. inhibiting compns., noncorrosive inhibitors and method for using)

IT 2236-96-2 2896-70-0, 4-Oxo-2,2,6,6-tetramethylpiperidine-1-oxyl

RL: MOA (Modifier or additive use); USES (Uses) (polymn. inhibitor; polymn. inhibiting compns., noncorrosive inhibitors and method for using)

RE.CNT 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

- (1) Anon; US 5856562 A CAPLUS
- (2) Anon; EP 810196 A1 CAPLUS
- (3) Japan Synthetic Rubber Co, Ltd; JP 05-202256 A 1993 CAPLUS
- (4) Mitsubishi Chemical Corp; JP 09-316026 A 1997 CAPLUS

L2 ANSWER 6 OF 44 CAPLUS COPYRIGHT 2002 ACS

AN 1999:78624 CAPLUS

DN 130:197965

TI Unsaturated polyester-based coating compositions containing N-substituted N-nitrosohydroxylamine salts

IN Yonezawa, Miwako; Yamazaki, Takahide

PA Nippon Shokubai Kagaku Kogyo Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 6 pp.

CODEN: JKXXAF

DT Patent
LA Japanese
IC ICM C09D167-06
ICS C09D005-04

CC 42-10 (Coatings, Inks, and Related Products)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 11029740	A2	19990202	JP 1997-186801	19970711
JP 3259402		B2	20020225		
AB	Coating compns. with good storage stability contain unsatd. polyesters, thixotropic agents and/or pigments, accelerators, and the title salts. Thus, a compn. contg. neopentyl glycol-propylene glycol-isophthalic acid-maleic anhydride-styrene copolymer 100, Aerosil 200 2, Co naphthenate 0.5, and N-nitroso-N-phenylhydroxylamine Al salt 0.015 part showed a small change in gelling time with Kayamec BUY after standing.				
ST	unsatd polyester coating nitroso hydroxylamine additive; thixotropic agent unsatd polyester nitrosohydroxylamine coating; glycol isophthalic maleic styrene copolymer coating; phthalic maleic styrene glycol copolymer coating				
IT	Naphthenic acids, uses RL: CAT (Catalyst use); USES (Uses) (cobalt salts, accelerator; unsatd. polyester-based coating compns. contg. nitrosohydroxylamine salts)				
IT	Coating materials (storage-stable; unsatd. polyester-based coating compns. contg. nitrosohydroxylamine salts)				
IT	Crosslinking catalysts Pigments, nonbiological Thixotropic agents (unsatd. polyester-based coating compns. contg. nitrosohydroxylamine salts)				
IT	Polyesters, uses RL: POF (Polymer in formulation); PRP (Properties); TEM (Technical or engineered material use); USES (Uses) (unsatd.; unsatd. polyester-based coating compns. contg. nitrosohydroxylamine salts)				
IT	13463-67-7, CR 90, uses RL: TEM (Technical or engineered material use); USES (Uses) (CR 90, pigment; unsatd. polyester-based coating compns. contg. nitrosohydroxylamine salts)				
IT	60650-95-5, Titanium yellow RL: TEM (Technical or engineered material use); USES (Uses) (TY 55, pigment; unsatd. polyester-based coating compns. contg. nitrosohydroxylamine salts)				
IT	7631-86-9, Aerosil 200, uses RL: MOA (Modifier or additive use); USES (Uses) (colloidal, thixotropic agent; unsatd. polyester-based coating compns. contg. nitrosohydroxylamine salts)				
IT	67939-21-3P, Isophthalic acid-maleic anhydride-neopentyl glycol-propylene glycol-styrene copolymer 220460-53-7P, Hexahydrophthalic anhydride-maleic anhydride-neopentyl glycol-propylene glycol-styrene copolymer RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)				

(unsatd. polyester-based coating compns. contg. nitrosohydroxylamine salts)

IT 135-20-6, **N-Nitroso-N-phenylhydroxylamine** ammonium salt 120457-86-5, **N-Nitroso-N-phenylhydroxylamine** aluminum salt
PL: MOA (Modifier or additive use); USES (Uses)
(unsatd. polyester-based coating compns. contg. nitrosohydroxylamine salts)

Page 44 CARMUS COPYRIGHT 2002 ACS

L2 ANSWER 10 OF 44 CAPS

AN 1992:591354

DN 117:191354 Isolation of N,N-dimethylacrylamide or N,N-dimethylmethacrylamide. Ito, Kido, Osamu

TI Preparation of N,N-dimethyl-*N*-[2-(*tert*-butylsulfonyl)ethyl]benzylamine. *Takashi* Hiraoka, Ryoichi; Okidaka, Isao; Kido, Susumu

IN Maruyama, Takashi, Kita-
ku, Tokyo, Japan

PA Kohjin Co., Ltd., Japan
5-10-12, Kokai, Tokkyo Koho, 6 pp.

58

CODEN: JN
B-100-1

DT Patent

LA Japanese
LA ICM 607C333-09

IC TCM C07C039-01
ICCS C07C231-12

TCS C-23-18 (Aliphatic Compounds)				APPLICATION NO.	DATE
FAN.CNT	1	PATENT NO.	KIND	DATE	-----
PI		JP 04154749	A2	19920527	JP 1990-276445 19901017
PI		JP 2986891	B2	19991206	
AB		N,N-Dimethyl(meth)acrylamide is prepd. by thermal decompn. of Me ₂ NCH ₂ CH ₂ CONMe ₂ (I; R = H, Me) in liq.-phase using vapor-phase polymn. inhibitors. Thermal decompn. of a mixt. of 3120 g I (R = H) and N -nitroso- N -phenylhydroxylamine ammonium salt			
ST		at 150-155.degree. for 30 h gave 2208 g crude monomers, which was vacuum distd. with phenothiazine under 10 mmHg to give 1525 g distd. with phenothiazine under 10 mmHg to give 1525 g N,N-dimethylacrylamide of 99.2% purity.			
IT		methylacrylamide prepn; methylmethacrylamide prepn; acrylamide dimethyl prepn; methacrylamide dimethyl prepn; thermal decompn dimethylaminoamid polymn inhibitor			
IT		Polymerization inhibitors (vapor-phase, in liq.-phase thermal decompn. of dimethylaminoamides)			
IT		17268-47-2	38872-39-8		
IT		RL: FCT (Reactant) (liq.-phase thermal decompn. of, vapor-phase polymn. inhibitors in)			
IT		10102-43-9, Nitrogen monoxide, uses			
IT		RL: USES (Uses) (polymn. inhibitor, in liq.-phase thermal decompn. of dimethylaminoamides)			
IT		135-20-6	3316-09-4	143814-78-2, Diphenylpicrylhydrazide	
IT		RL: RCT (Reactant) (polymn. inhibitor, in liq.-phase thermal decompn. of dimethylaminoamides)			
IT		3680-03-7P, N,N-Dimethylacrylamide			
IT		Dimethylmethacrylamide RL: SPN (Synthetic preparation); PREP (Preparation) (prepns. of)			

15 OF 44 GABRIUS COPYRIGHT 2002 ACS

L2 ANSWER 15 OF 44 CA
20501 GABRIEL

AN 1988:38591

TI Polymerization inhibitors for acrylic monomers

IN Mukoyama, Hideaki; Hiraoka, Ryochi

PA Kureha Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC C08F020-56

ICS C07C081-00; C08F002-00

CC 35-2 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 23

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
------------	------	------	-----------------	------

PI JP 62187710 A2 19870817 JP 1986-28911 19860214

AB Polymn. inhibitors for H2C:C(R)CONHR1 (I; R = H, methyl; R1 = C1-3 alkyl) comprise nitroso compds., inorg. stabilized radicals, and/or org. stabilized radicals. Thus, when 10 parts I (R = H, R1 = Me) was heated

at 100.degree. under reduced pressure for 23 h in the presence of 0.05 part 4-ONC6H4NMe2, no polymn. was obsd. even after 20 h continuous heating at 130.degree..

ST polymn inhibitor alkyl acrylamide; nitroso compd polymn inhibitor; methacrylamide alkyl polymn inhibitor; nitrosodimethylaniline polymn inhibitor methylacrylamide

IT Nitroso compounds

PL: USES (Uses)
(polymn. inhibitors, for alkyl(meth)acrylamides)

IT Polymerization inhibitors
(popcorn, for alkyl(meth)acrylamides, nitroso compds. or (in)org. radicals as)

IT 1187-59-3, N-Methylacrylamide 3887-02-3, N-Methylmethacrylamide
25999-13-7, N-Propylacrylamide

PL: USES (Uses)
(polymn. inhibitor for, nitroso compds. or (in)org. radicals as)

IT 131-91-9, 1-Nitroso-2-hydroxynaphthalene 132-53-6, 2-Nitroso-1-hydroxynaphthalene 135-20-6, **N-Nitroso-N-phenylhydroxylamine** ammonium salt 138-89-6, N,N-Dimethyl-4-nitrosoaniline 586-96-9, Nitrosobenzene 2370-18-5, Galvinoxyl 2564-83-2 2896-70-0 10102-43-9, uses and miscellaneous 10102-44-0, uses and miscellaneous 24973-59-9, 2,4,6-Tri-tert-butylnitrosobenzene 30772-85-1, Nitrosodiphenylphenylamine 112340-28-0

PL: USES (Uses)
(polymn. inhibitors, for alkyl(meth)acrylamides)

L2 ANSWER 26 OF 44 CAPLUS COPYRIGHT 2002 ACS

AN 1973:16804 CAPLUS

NN 78:16804

TI Polymerization inhibitors for the catalytic hydration of acrylonitrile

IN Modeen, James H.; Newton, Gary E.

PA Dow Chemical Co.

SO U.S., 3 pp.

CODEN: USXXAM

DT Patent

LA English

IC C07C

NCL 260561000N

CC 35-4 (Synthetic High Polymers)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	US 3689558	A	19720905	US 1970-17943	19700309
	BE 771665	A1	19720223	BE 1971-107349	19710823
PRAI	US 1970-17943		19700309		
AB	Polymer formation during the catalytic hydration of acrylonitrile (I) to acrylamide (II) was reduced or prevented by the addn. of N-nitroso-N-phenylhydroxylamine ammonium salt (III) [135-20-6], a nitrosophenol, or a trialkylamine contg. Cl-6 alkyl groups where < 2 alkyl groups were Me. The inhibitors could be deactivated by adjusting the pH so that the II could be polymerized. Thus, a neutral 7% I soln. contg. 25 ppm III was heated at 90.deg. in the presence of reduced catalyst contg. 40% Cu and 25.5% Cr to yield II without polymer formation. The II was made basic and was polymerized in the presence of a persulfate initiator to form polyacrylamide				
[9003-05-8], which was as good as the polymer obtained from II produced without an inhibitor. I hydrated over the reduced catalyst without the inhibitor or in the presence of hydroquinone Me ether formed polymer in the catalyst and in the reactor. Triethylamine [121-44-8] was also used instead of III, and produced similar polymer-free II.					
ST	acrylonitrile polymn inhibitor ethylamine; nitrosophenylhydroxylamine polymn inhibitor; amine polymn inhibitor acrylonitrile; acrylamide sepn acrylonitrile hydration				
IT	Polymerization inhibitors (nitrosophenylhydroxylamine ammonium salt, for acrylamide)				
IT	Hydration, chemical (of acrylonitrile, to acrylamide)				
IT	79-06-1P, preparation RL: PREP (Preparation) (from acrylonitrile, by hydration)				
IT	11104-65-7 39320-46-2 RL: USES (Uses) (hydration of acrylonitrile to acrylamide in presence of)				
IT	107-13-1, reactions FL: RCT (Peactant) (hydration of, to acrylamide)				
IT	121-44-8, uses and miscellaneous 135-20-6 RL: USES (Uses) (inhibitors, for polymn. of acrylamide)				

L2 ANSWER 33 OF 44 CAPLUS COPYRIGHT 2002 ACS

AN 1966:457247 CAPLUS

DN 65:57247

OREF 65:10697f, 10698a

TI Polymerization inhibitors
PA Copolymer Rubber and Chemical Corp.

SO 22 pp.

DT Patent

LA Unavailable

IC C08F

CC 45 (Synthetic High Polymers)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	NL 65011747		19660311	NL	
PRAI	US		19640910		

AB Org. N-nitrosohydroxylamines or salts thereof are used in vinylpyridine, unsatd. hydrocarbons, and unsatd. esters to inhibit thermal polymerization and (or) the growth of popcorn polymer therein, e.g. during storage.

They need not be removed prior to use of the monomers in catalytic polymerizations. NH₄ salts of **N-nitroso-N-phenylhydroxylamine** (I) or of N-nitroso-N-(1-naphthyl)hydroxylamine are very useful. I is twice as effective as a thermal polymerization inhibitor than the conventional tert-butyldiphenylpyrocatechol (II). Copolymerization of butadiene with styrene by a cold rubber polymerization process is not retarded by I while II prevents any reaction.

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=> s 148-97-0/rn
      79 148-97-0
      30 148-97-0D
L6      54 148-97-0/RN
      (148-97-0 (NOTL) 148-97-0D )
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IND ----- Indexing data
IPC ----- International Patent Classifications
MAX ----- ALL, plus Patent FAM, RE
FATS ----- PI, SO
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e.g., D SCAN or DISPLAY SCAN)
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IALL ----- ALL, indented with text labels
IBIB ----- BIB, indented with text labels
IMAX ----- MAX, indented with text labels
ISTD ----- STD, indented with text labels

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OIBIB ----- OBIB, indented with text labels

SBIB ----- BIB, no citations
SIBIB ----- IBIB, no citations

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HITSEQ ----- HIT FN, its text modification, its CA index name, its structure diagram, plus NTE and SEQ fields
FHITSTR ----- First HIT FN, its text modification, its CA index name, and its structure diagram
FHITSEQ ----- First HIT FN, its text modification, its CA index name, its structure diagram, plus NTE and SEQ fields
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OCC ----- Number of occurrence of hit term and field in which it occurs

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	L6	ANSWER 1 OF 54	CAPLUS	COPYRIGHT 2002 ACS	
	AN	2002:353510	CAPLUS		
	DN	136:370166			
	TI	Irradiation process for making olefin graft copolymers with low molecular-weight side chains			
	IN	Dang, Vu A.; Phan, Tam T. M.; Smith, Jeanine A.; Song, Cheng Q.			
	PA	Basell Technology Company B.V., Neth.			
	SO	PCT Int. Appl., 23 pp.			
	DT	CODEN: PIKKDR			
	LA	Patent			
	FAN.CNT	English			
	1				
	PATENT NO.	KIND	DATE	APPLICATION NO.	
	-----	-----	-----	-----	
PI	WO 2002036650	A1	20020510	WO 2001-IB2014	
	W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BF, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TF, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KE, MD, RU, TJ, TM		20011026	
	RW:	GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BT, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MF, NE, SN, TD, TG			
	PRAI US	2000-704375	A	20001102	
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L1 0 S N-NITECS-N-PHENYLHYDROXYLAMINE
L2 44 S N-NITROSO-N-PHENYLHYDROXYLAMINE
L3 803 S 100-65-2/RN
L4 0 S L2 AND L3

FILE 'STNGUIDE' ENTERED AT 13:54:27 ON 29 JUN 2002
L5 0 S 148-97-0/RN

FILE 'CPLUS' ENTERED AT 14:01:43 ON 29 JUN 2002
L6 54 S 148-97-0/RN

=> s 3710-84-7/rn
739 3710-84-7
15 3710-84-7D
L7 726 3710-84-7/RN
(3710-84-7 (NOTL) 3710-84-7D)

=> s 16 and 17
L8 6 L6 AND L7

=> d 18 ti

L8 ANSWER 1 OF 6 CPLUS COPYRIGHT 2002 ACS
TI Irradiation process for making olefin graft copolymers with low
molecular-weight side chains

=> d 18 2-6 ti

L8 ANSWER 2 OF 6 CPLUS COPYRIGHT 2002 ACS
TI Preparation of polyolefin graft copolymers with low molecular weight side
chains by using polymeric peroxide as initiator

L8 ANSWER 3 OF 6 CPLUS COPYRIGHT 2002 ACS
TI Stabilized monomer composition

L8 ANSWER 4 OF 6 CPLUS COPYRIGHT 2002 ACS
TI Preparation of alicyclic epoxidated dihydroniclopentadienyl
(meth)acrylates as monomers

L8 ANSWER 5 OF 6 CPLUS COPYRIGHT 2002 ACS
TI Manufacture of epoxycyclohexylmethyl (meth)acrylates

L8 ANSWER 6 OF 6 CPLUS COPYRIGHT 2002 ACS
TI Epoxidation of cyclohexenylmethyl (meth)acrylate in presence of
polymerization inhibitors

=> d 18 6 ali

L8 ANSWER 6 OF 6 CPLUS COPYRIGHT 2002 ACS
AN 1990:632247 CPLUS

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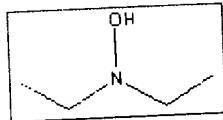
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diethyl hydroxylamine [3710-84-7]

Synonyms: Ethanamine, N-ethyl-N-hydroxy-; N,N-Diethylhydroxylamine;

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89.137

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[Density](#)

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113:232247

L8 ANSWER 5 OF 6 CAPLUS COPYRIGHT 2002 ACS
AN 1990:632247 CAPLUS
DN 113:232247
TI Epoxidation of cyclohexenylmethyl (meth)acrylate in presence of polymerization inhibitors
IN Fukuya, Kazuaki; Kuwana, Akihiro
PA Daicel Chemical Industries, Ltd., Japan
SO Jpn. Kokai Tokkyo Kcho, 9 pp.
C0DEN: JKXKAF
PT Patent
LA Japanese
IC ICM C07D301-36
ICS C07D303-16; C08F020-32; C08G059-20; C09D004-02; C09D133-06;
C09D163-00
CC 35-2 (Chemistry of Synthetic High Polymers)
Section cross-reference(s): 27
FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 02188576	A2	19900724	JP 1989-5816	19890112
JP 2704284	B2	19980126		

OS MARPAT 113:232247
AB 3,4-Epoxycyclohexenylmethyl acrylate and methacrylate (I) are prep'd. by epoxidn. of 3-cyclohexen-1-ylmethyl acrylate and methacrylate (II) with an oxidizing agent in the presence of polymn. inhibitors comprising .gtoreq.1 compd. selected from hydroquinone, hydroquinone mono-Me ether (III), p-benzoquinone, cresol, tert-butylcatechol, phenols substituted by tert-Bu and other groups, 2,5-dihydroxy-p-quinone, piperidine, ethanolamine, .alpha.-nitroso-.beta.-naphthol, HNPh₂, phenothiazine, N-nitrosophenylhydroxylamine, and Et₂NOH and .gtoreq.1 compd. selected from H₃PO₄, K₃PO₄, Na₃PO₄, Na(NH₄)HPO₄, H₄P2O₇, K₄P2O₇, Na₄P2O₇, 2-ethylhexyl pyrophosphate, K or Na 2-ethylhexyl pyrophosphate, tripolyphosphoric acid, K or Na 2-ethylhexyl tripolyphosphate, and Na or K 2-ethylhexyl tetrapolyphosphate. Thus, a mixt. of 14.4 kg II, 52.8 kg AcOEt, 1 g III, and 12 g H₄P2O₇ was treated with 24.8 kg 30% AcOOH during 4 h at 50-degree. and aged 4 h to give 14.2 kg product contg. 94.7% I, 1 g of which dissolved completely in 10 g heptane.
ST epoxycyclohexenylmethyl acrylate prep'n polymn inhibitor; methacrylate epoxycyclohexenylmethyl prep'n polymn inhibitor; epoxidn cyclohexenylmethyl acrylate polymn inhibitor; hydroquinone polymn inhibitor methacrylate; pyrophosphoric polymn inhibitor acrylate; phenol polymn inhibitor acrylate; amine polymn inhibitor acrylate; phosphate polymn inhibitor acrylate
IT Polymerisation inhibitors
(in epoxidn. of cyclohexenylmethyl (meth)acrylate)
IT Epoxidation
(of cyclohexenylmethyl (meth)acrylate, polymn. inhibitors in)
IT Phenols, uses and miscellaneous
EL: WSES (Uses)
(polymn. inhibitors, in epoxidn. of cyclohexenylmethyl (meth)acrylate)
IT 21367-03-3, 3-Cyclohexen-1-ylmethyl acrylate 21367-03-3,

3-Cyclohexen-1-ylmethyl methacrylate

FL: RCT (Reactant)

(epoxidn. of, polymn. inhibitors in)

IT 88-32-4, 3-tert-Butyl-4-methoxyphenol 92-84-2, Phenothiazine

106-51-4,

p-Benzoquinone, uses and miscellaneous 110-89-4, Piperidine, uses and miscellaneous 121-00-6, 2-tert-Butyl-4-methoxyphenol 122-39-4, Diphenylamine, uses and miscellaneous 123-31-9, Hydroquinone, uses and miscellaneous 128-37-0, 2,6-Di-tert-butyl-p-cresol, uses and miscellaneous 131-91-9, .alpha.-Nitroso-.beta.-naphthol 141-43-5, Ethanolamine, uses and miscellaneous 148-97-0 150-76-5, Hydroquinone monomethyl ether 615-94-1 1319-77-3, Cresol 1693-78-3, 2-Ethylhexyl pyrophosphate 1879-09-0, 2,4-Dimethyl-6-tert-butylphenol 2466-09-3, Pyrophosphoric acid 3710-84-7, N,N-Diethylhydroxylamine 7320-34-5, Potassium pyrophosphate 7632-05-5, Sodium phosphate 7664-38-2, Phosphoric acid, uses and miscellaneous 7722-88-5 10380-08-2, Tripolyphosphoric acid 12767-83-8, Sodium 2-ethylhexyl tripolyphosphate 13011-54-6, Ammonium sodium hydrogen phosphate 16068-46-5, Potassium phosphate 27213-78-1, tert-Butylcatechol 130455-01-5 130455-02-6 130455-03-7

130455-55-1

130455-66-2

RL: USES (Uses)

(polymn. inhibitors, in epoxidn. of cyclohexenylmethyl (meth)acrylate)

IT 64630-63-3P, 3,4-Epoxyhexylmethyl acrylate 82428-30-6P,
3,4-Epoxyhexylmethyl methacrylate

RL: PREP (Preparation)

(prepn. of, by epoxidn., polymn. inhibitors in)